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Summary:

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Quantitative Chemical Analysis - web.chem.ucsb.edu 3 LE CHATELIER'S PRINCIPLE $\text{NaCl(s)} + \text{H}_2\text{SO}_4\text{(l)} \rightleftharpoons \text{Na}_2\text{SO}_4\text{(s)} + \text{HCl(g)}$ reaction is carried out in an open container Because HCl is a gas that can escape from the system.

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